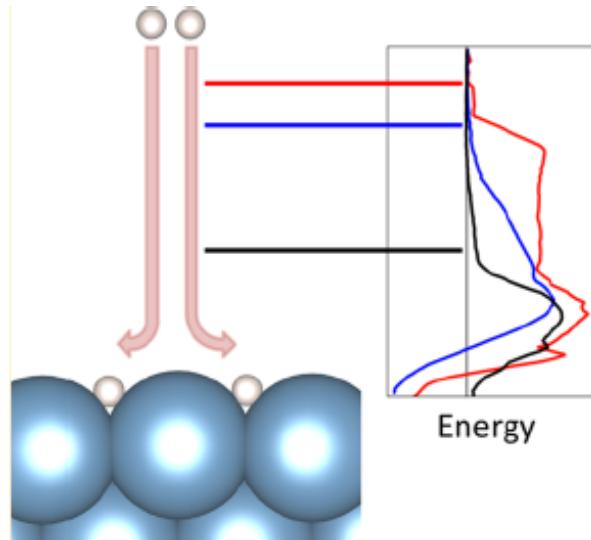
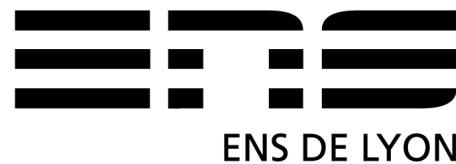
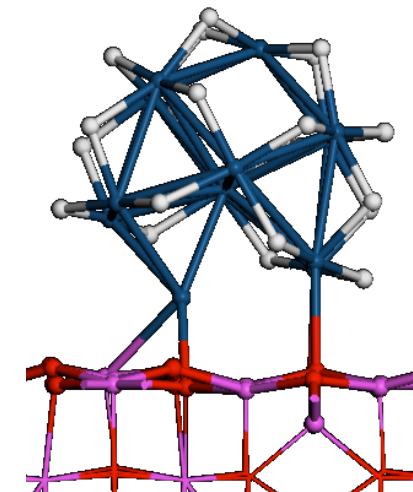


# Modelling heterogeneous catalysis: what challenge for first principle calculations?



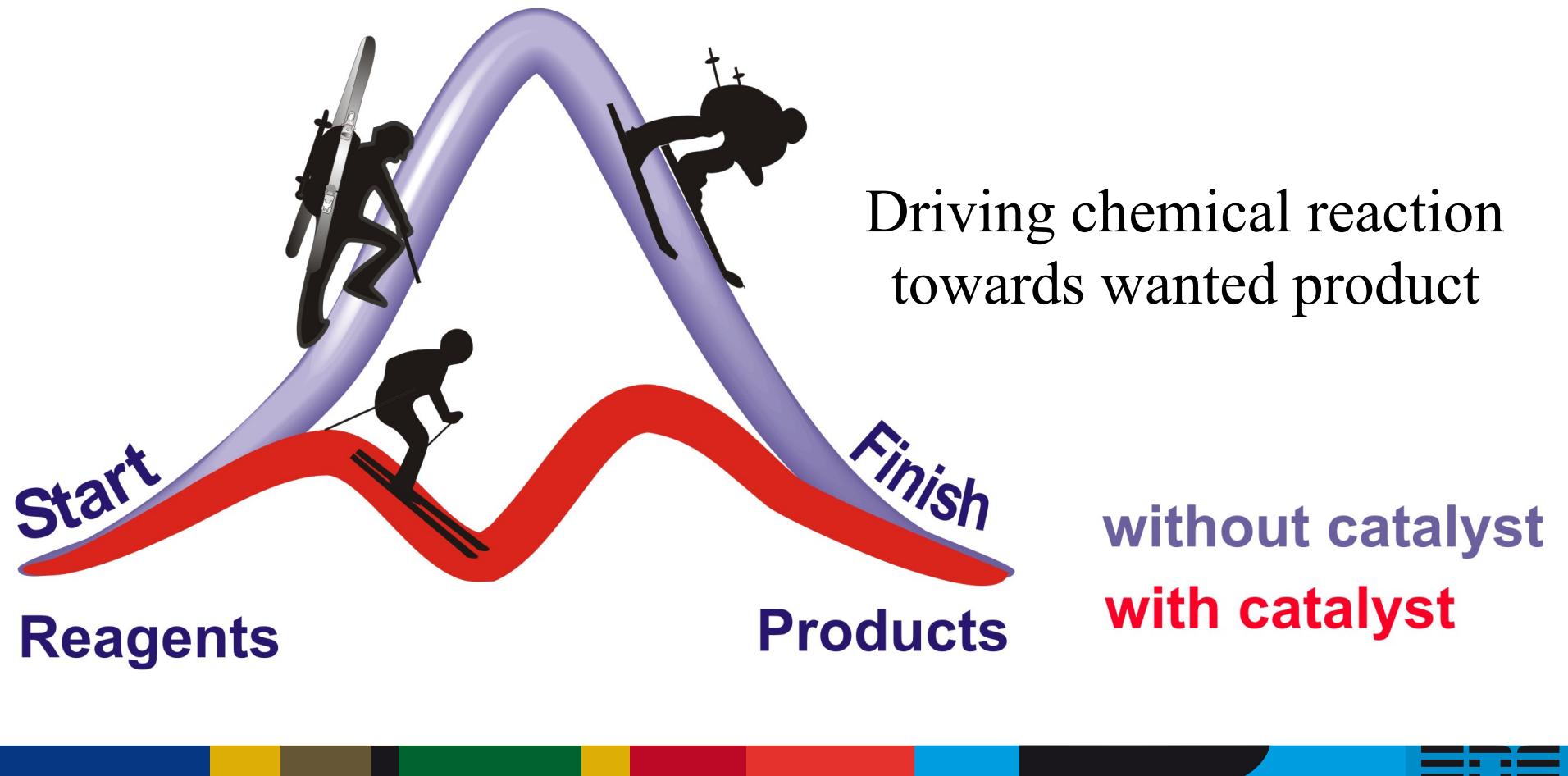
Philippe Sautet

Institute of Chemistry  
University of Lyon



# Making chemical reactions easier with catalysis

Lowering the energy barrier, finding an easy path



# Ecoefficient chemistry with catalysis ... and simulations

Chemical plant of the 21<sup>st</sup> century

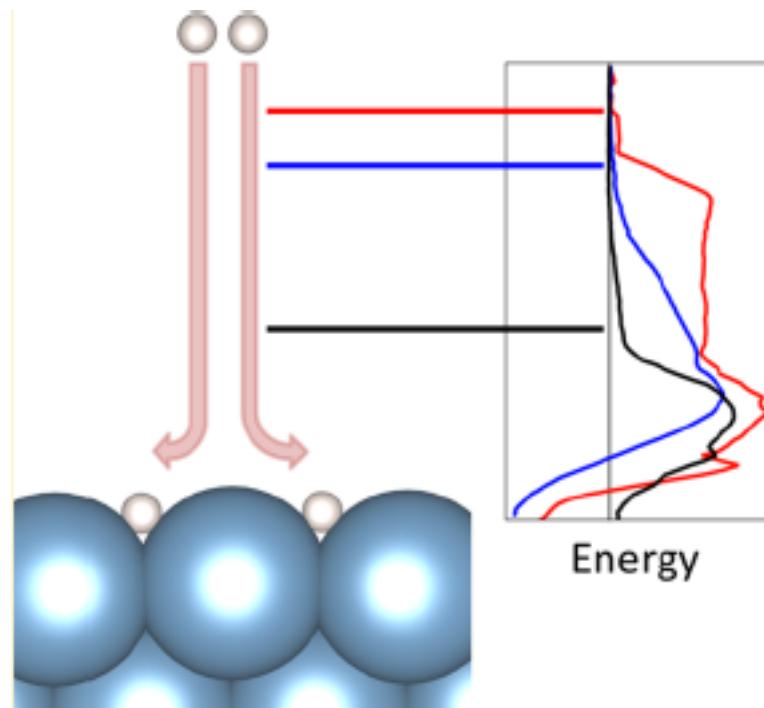


- Soft conditions
- total selectivity
- No waste

Design of efficient catalysts  
Understanding mechanisms at the molecular scale  
Molecular simulation is a key approach

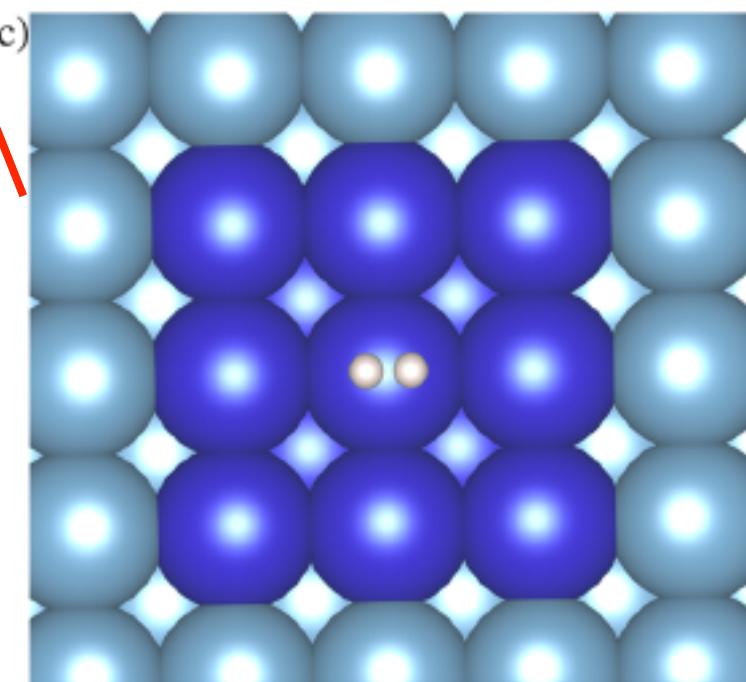


# Reaction pathways at surfaces

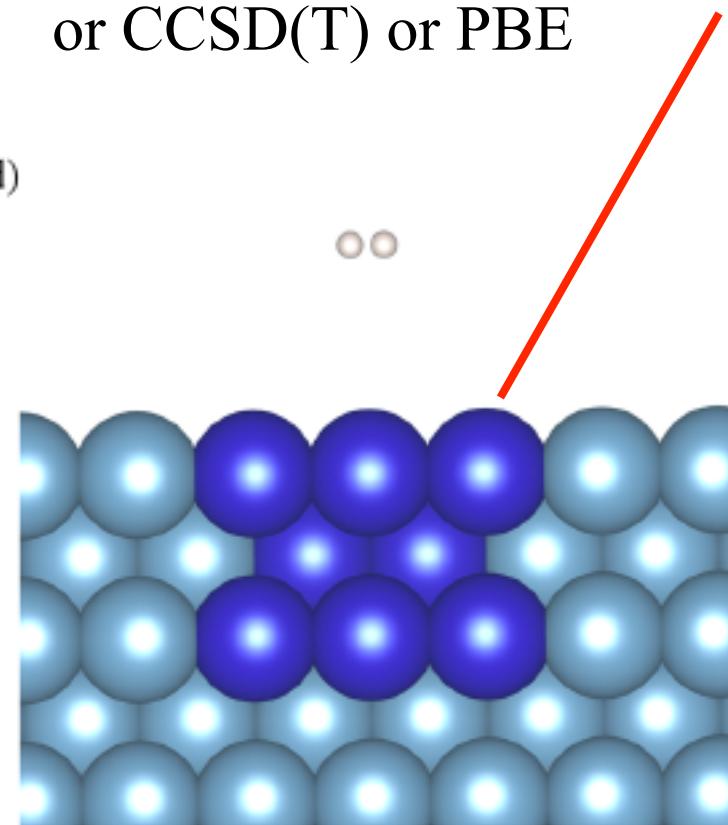


# $H_2$ on Cu(100): hybrid approach

5 layers 3x3 Cu (100) surface: PBE



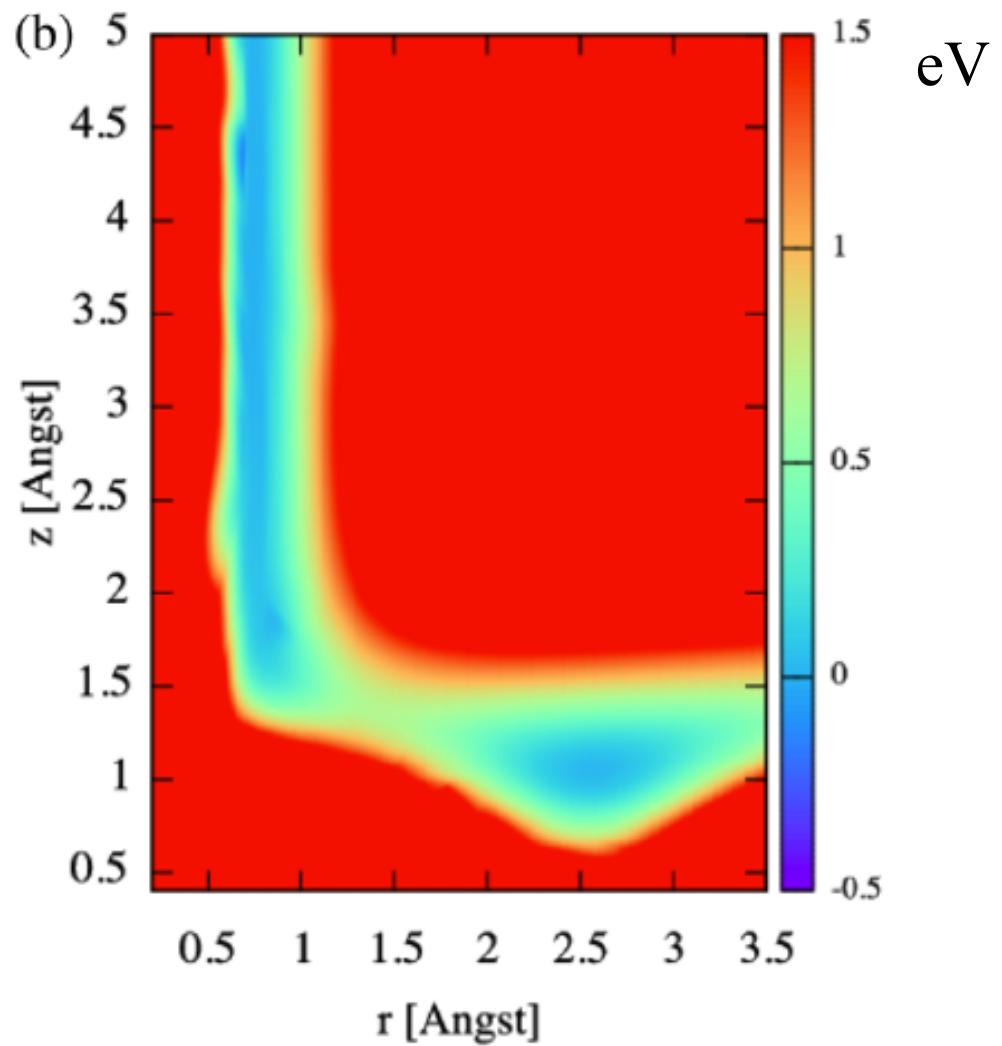
Cu<sub>22</sub> cluster : MRCI+Q  
or CCSD(T) or PBE



ONIOM type embedding



# Potential energy surface

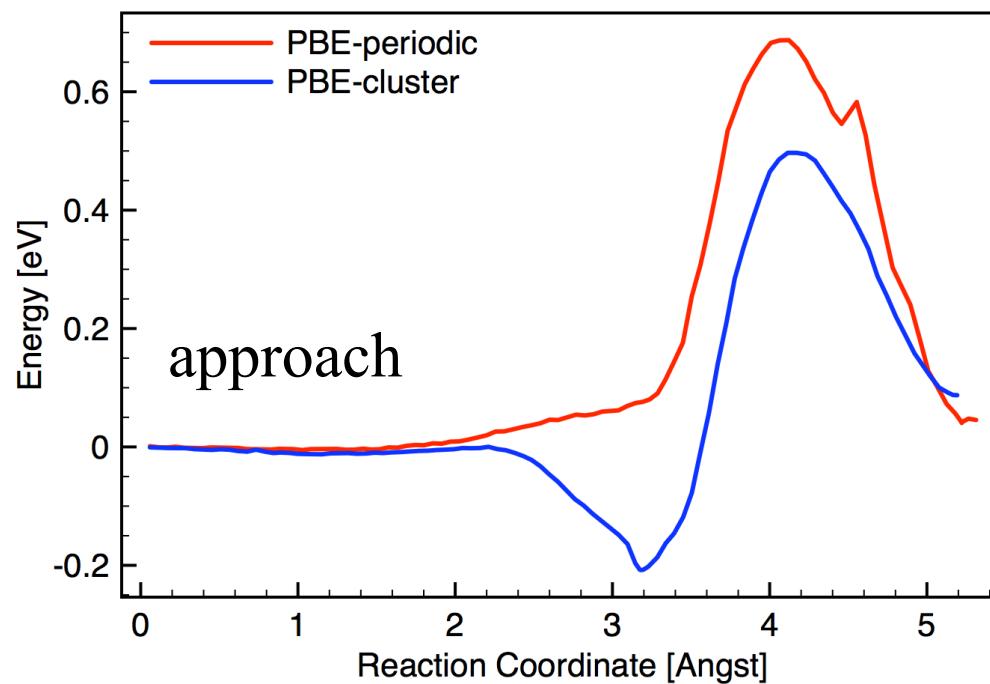


# PBE/PBE

TS  $r=1.48 \text{ \AA}$

$Z=1.33 \text{ \AA}$

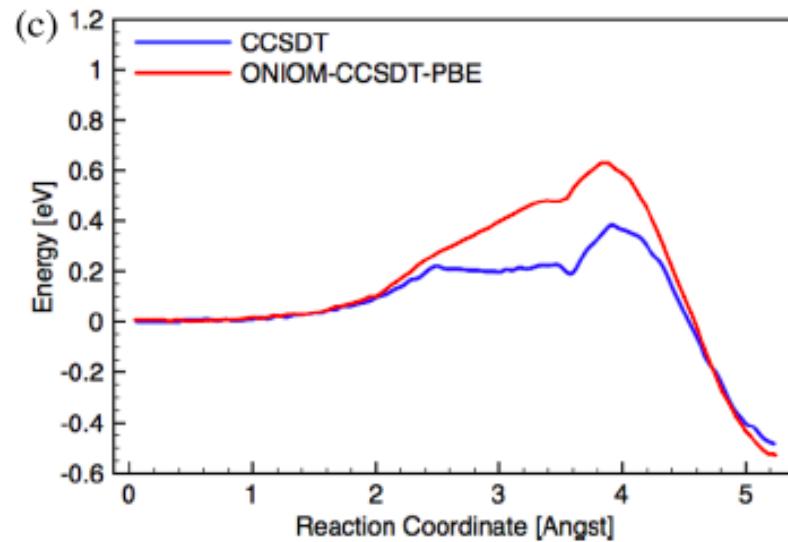
$E^{\text{TS}}=0.69 \text{ eV}$



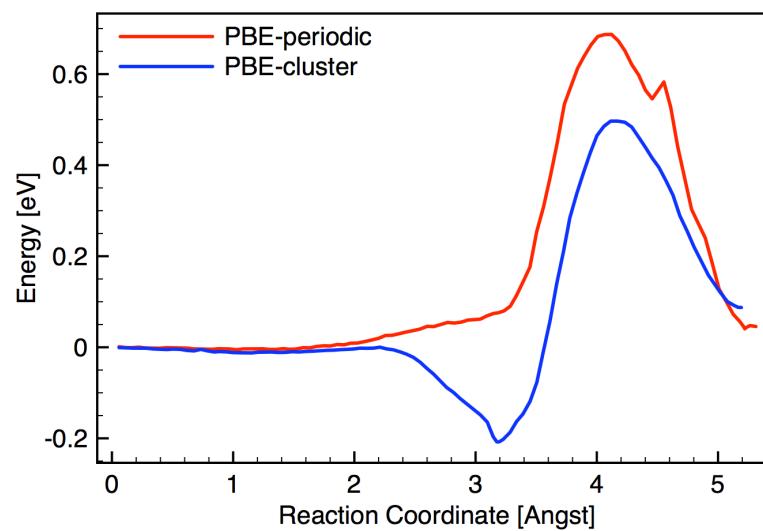
min  $r=2.49 \text{ \AA}$   
 $Z=1.01 \text{ \AA}$



# CCSD(T)/PBE



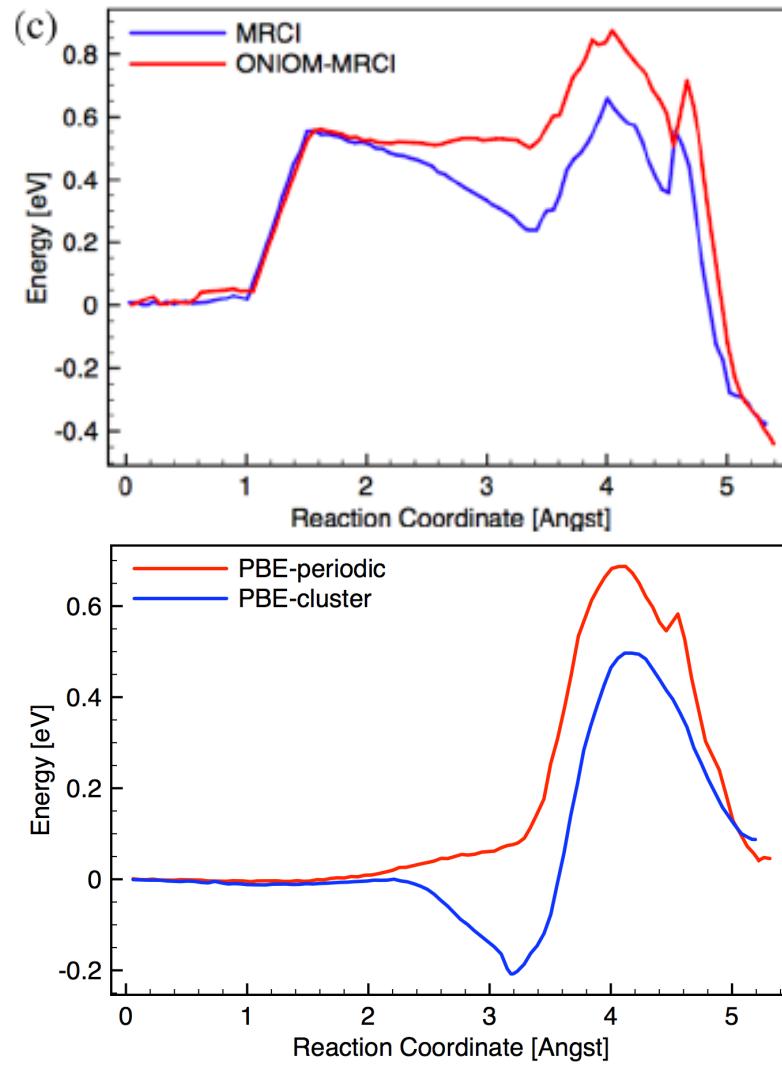
TS r=1.41 Å  
Z=1.19 Å  
 $E^{TS}=0.63$  eV



TS r=1.48 Å  
Z=1.33 Å  
 $E^{TS}=0.69$  eV



# MRCI+Q/PBE



TS  $r=1.39 \text{ \AA}$   
 $Z=1.29 \text{ \AA}$   
 $E^{TS}=0.87 \text{ eV}$

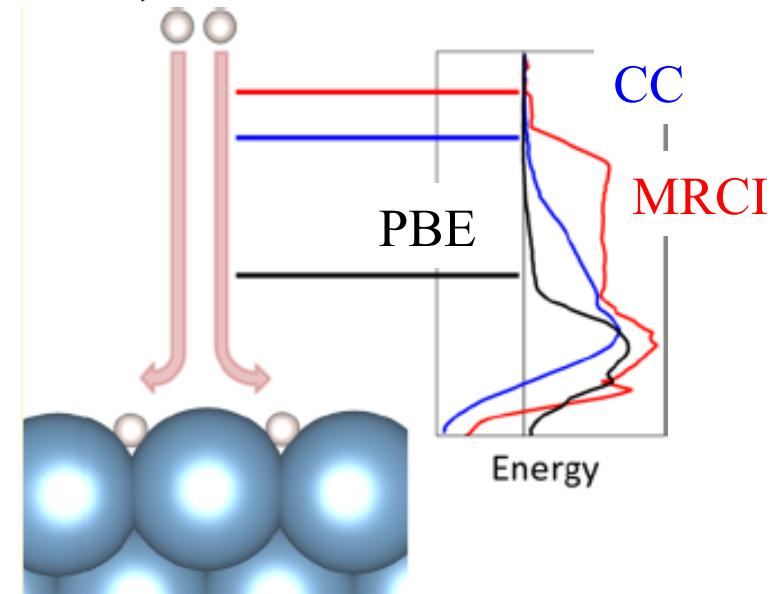
TS  $r=1.48 \text{ \AA}$   
 $Z=1.33 \text{ \AA}$   
 $E^{TS}=0.69 \text{ eV}$

Diaz et al. SRP-DFT  
 $E^{TS}=0.87 \text{ eV}$



# $H_2$ on Cu(100)

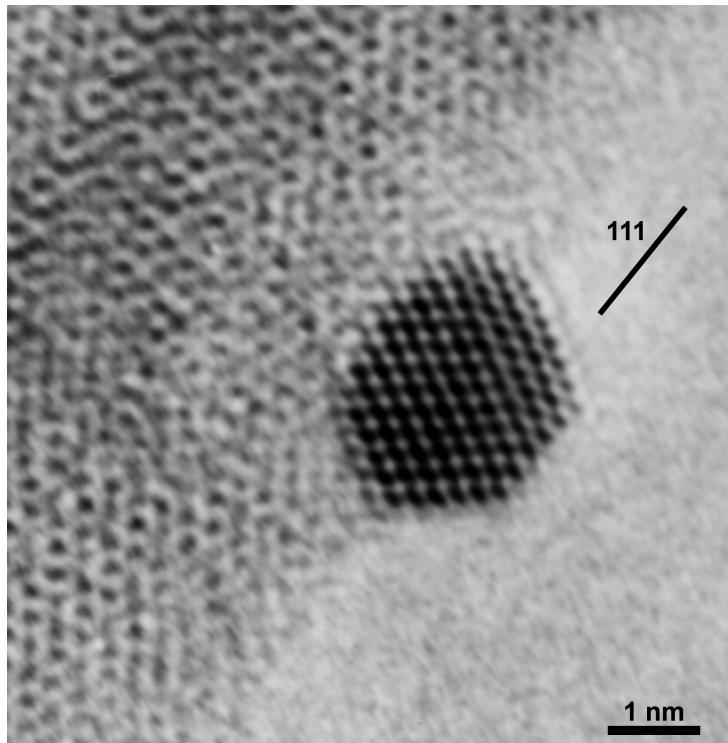
- Hybrid QM/QM' scheme
- Cluster energies need to be corrected
- Much broader barrier with explicitly correlated calculations
- Activation energy with MRCI+Q / PBE in excellent agreement with best estimate of experimental value
- DFT/PBE is (only) 0.18 eV away



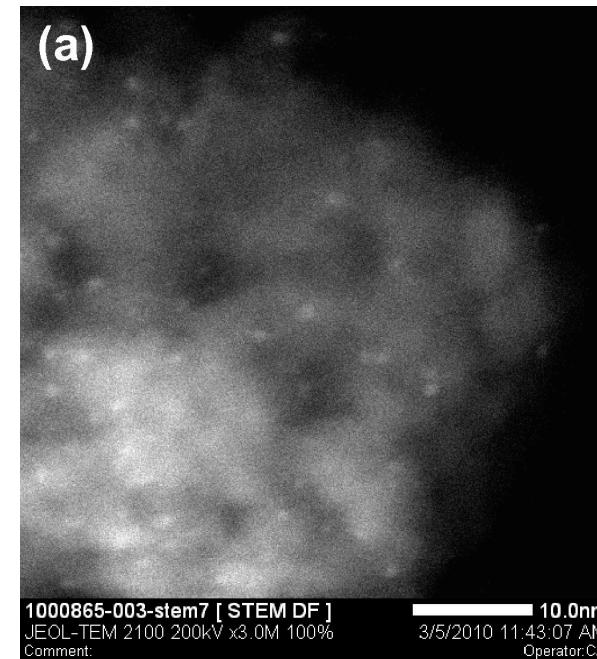
F. Görtl, C. Houriez, M. Guitou, G. Chambaud and P. Sautet J. Phys. Chem. C. 118, 5374-5382 (2014)



# Pt particles on $\gamma$ -alumina



Particle size 0.6 – 1.1 nm



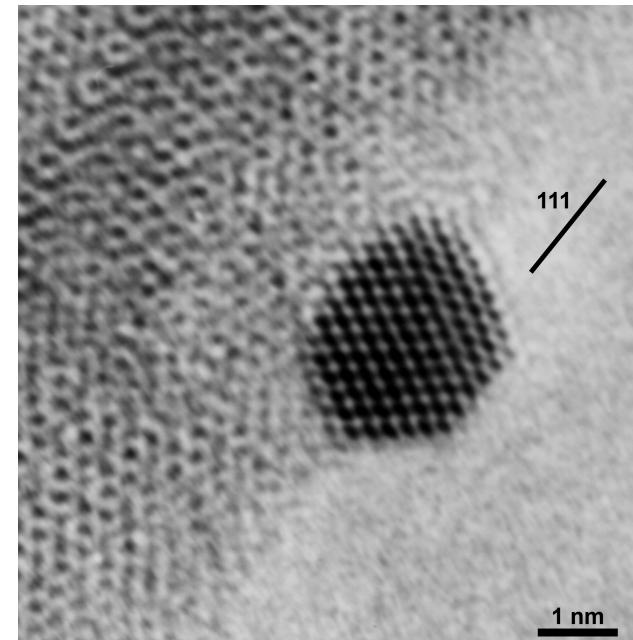
A. Jael et al, J. Catal. 272 (2010) 275

$\text{Pt}_{10} - \text{Pt}_{20}$



# Supported particles: open questions

- Influence of size
- Shape, various sites
- Influence of support  
Electronic transfer
- Specific chemisorption  
properties
- Catalytic reactivity

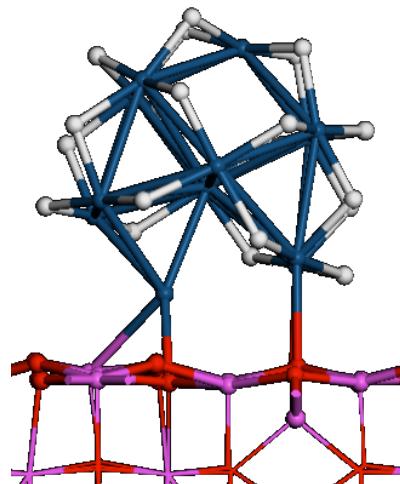


Nano-particle of Pt on Alumina

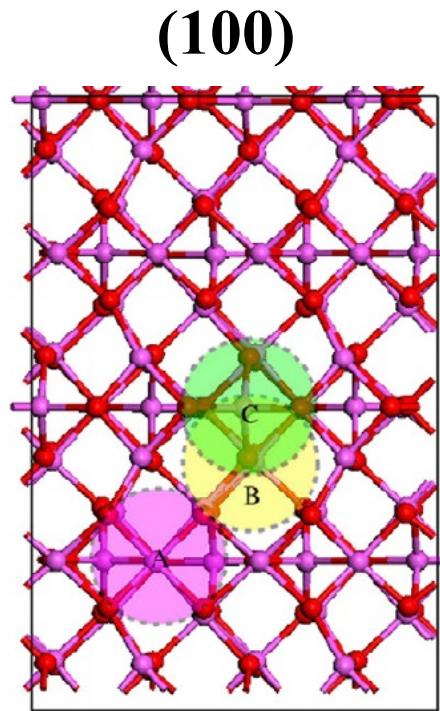


# Theoretical Methods

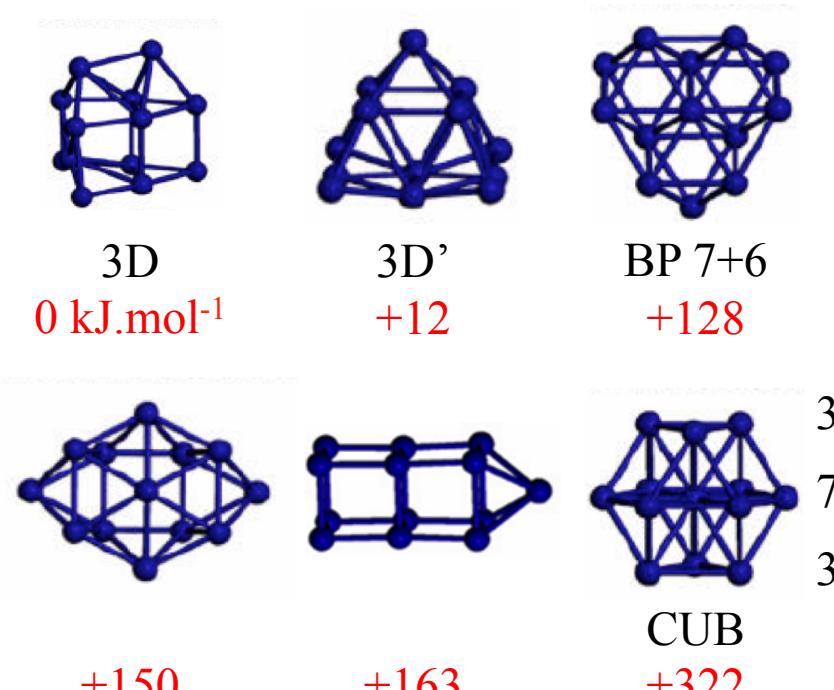
- Catalyst: nanoparticle deposited on extended support
- Density functional theory  
GGA: Perdew-Wang 91 or PBE
- Structural exploration with MD
- Combination with thermodynamics



# Pt<sub>13</sub> particles on the $\gamma$ -Al<sub>2</sub>O<sub>3</sub> support



relevant Pt<sub>13</sub> shapes



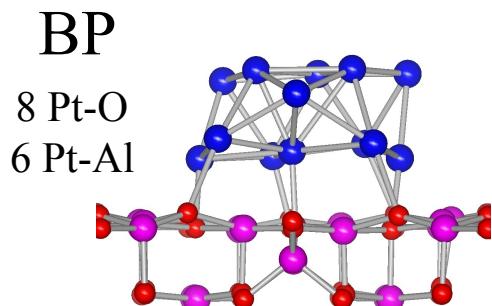
(100) is fully dehydrated

In vacuum

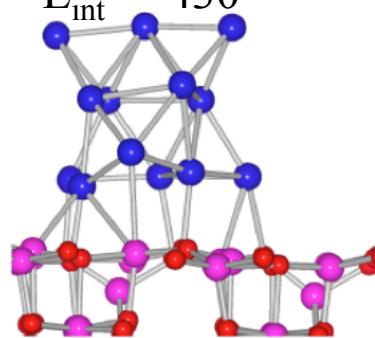


# Pt<sub>13</sub> particles on γ-Al<sub>2</sub>O<sub>3</sub> (100)

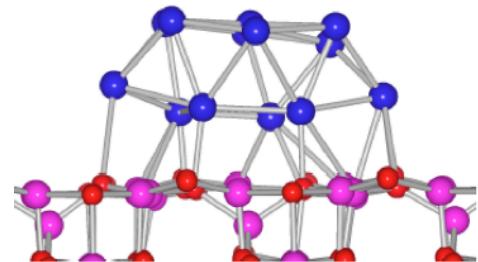
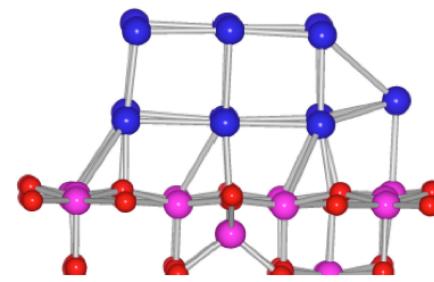
$E = 0 \text{ kJ.mol}^{-1}$   
 $(E_{\text{gas}} = +136)$   
 $E_{\text{int}} = -783$



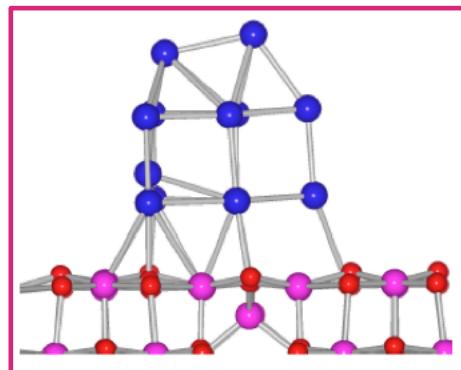
$E = +25$   
 $(E_{\text{gas}} = +12)$   
 $E_{\text{int}} = -430$



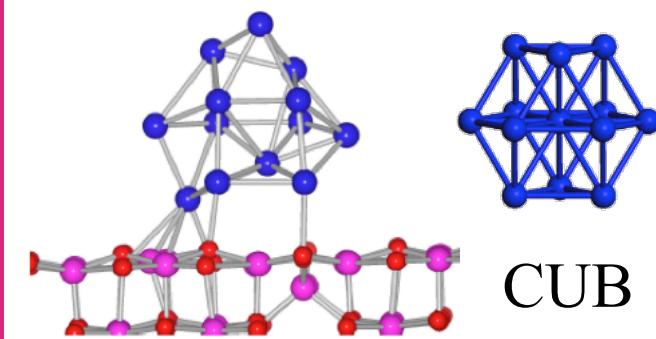
$E = +26$   
 $(E_{\text{gas}} = +163)$   
 $E_{\text{int}} = -704$



$E = +62$   
 $(E_{\text{gas}} = +150)$   
 $E_{\text{int}} = -646$



$E = +62$   
 $(E_{\text{gas}} = 0)$   
 $E_{\text{int}} = -300$



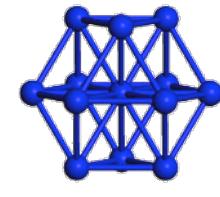
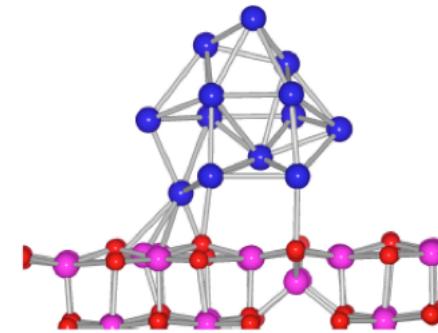
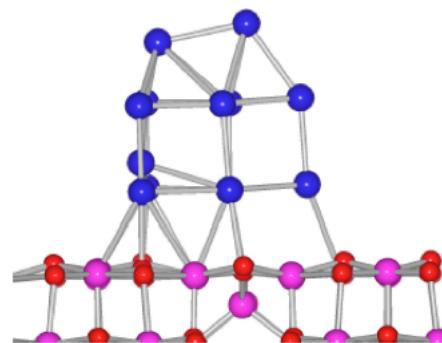
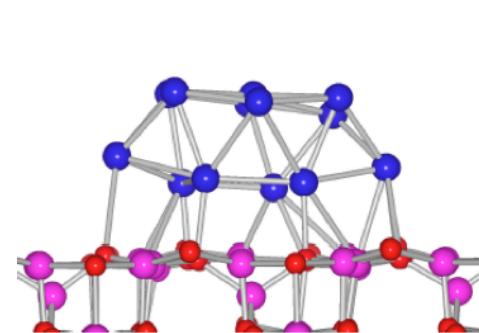
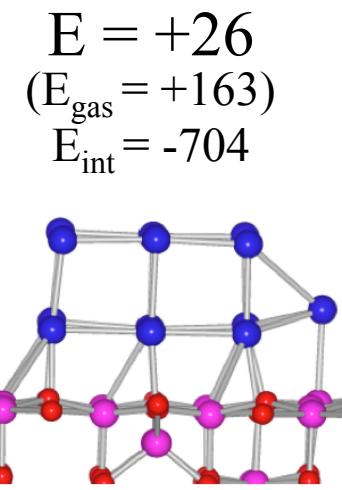
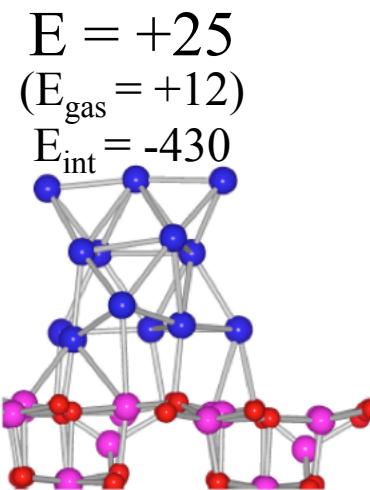
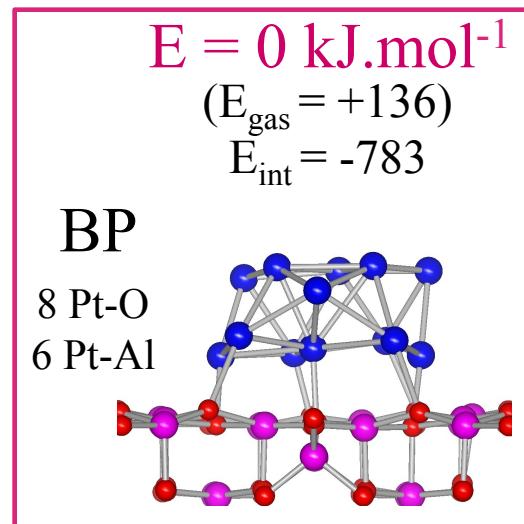
$E = +150$   
 $(E_{\text{gas}} = +322)$   
 $E_{\text{int}} = -331$



C-H. Hu, C. Chizallet, C. Mager-Maury, M. Corral-Valero, P. Sautet, H. Toulhoat and P. Raybaud, *Journal of Catalysis* 274, 99-110 (2010)



# Pt<sub>13</sub> particles on γ-Al<sub>2</sub>O<sub>3</sub> (100)



CUB

$E = +62$   
 $(E_{\text{gas}} = +150)$   
 $E_{\text{int}} = -646$

$E = +62$   
 $(E_{\text{gas}} = 0)$   
 $E_{\text{int}} = -300$

$E = +150$   
 $(E_{\text{gas}} = +322)$   
 $E_{\text{int}} = -331$

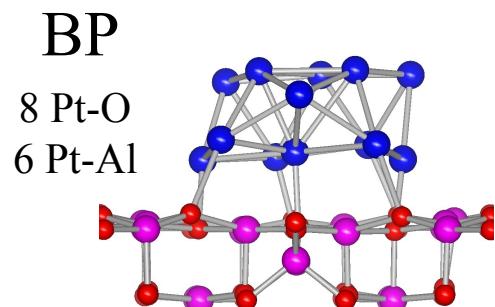


C-H. Hu, C. Chizallet, C. Mager-Maury, M. Corral-Valero, P. Sautet,  
H. Toulhoat and P. Raybaud, *Journal of Catalysis* 274, 99-110 (2010)

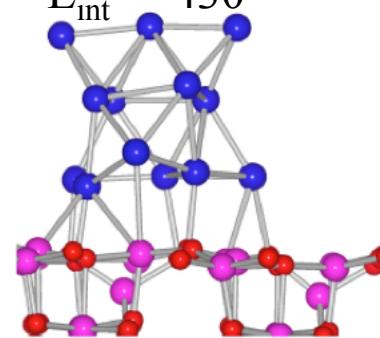


# Pt<sub>13</sub> particles on γ-Al<sub>2</sub>O<sub>3</sub> (100)

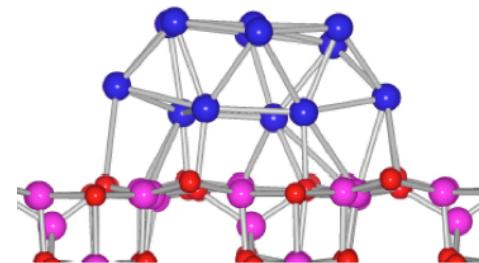
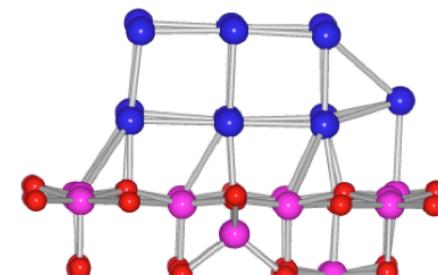
$E = 0 \text{ kJ.mol}^{-1}$   
 $(E_{\text{gas}} = +136)$   
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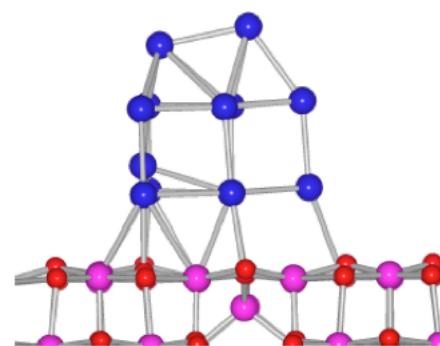
$E = +25$   
 $(E_{\text{gas}} = +12)$   
 $E_{\text{int}} = -430$



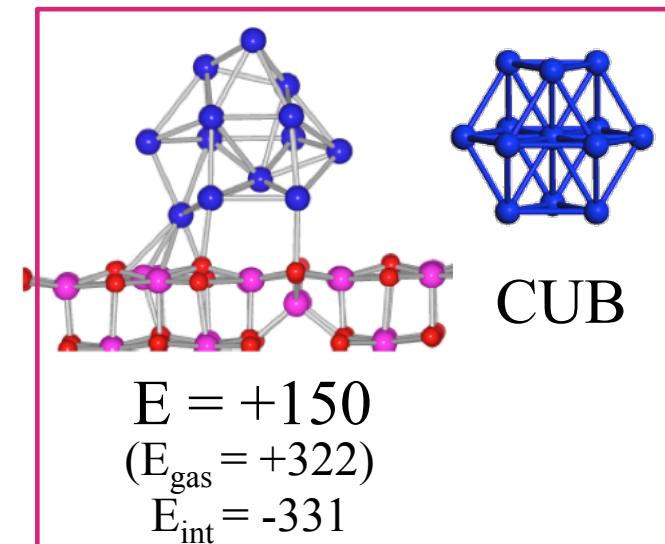
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$E = +62$   
 $(E_{\text{gas}} = +150)$   
 $E_{\text{int}} = -646$



$E = +62$   
 $(E_{\text{gas}} = 0)$   
 $E_{\text{int}} = -300$



● Al   ● O   ● Pt

C-H. Hu, C. Chizallet, C. Mager-Maury, M. Corral-Valero, P. Sautet, H. Toulhoat and P. Raybaud, *Journal of Catalysis* 274, 99-110 (2010)



Realistic  
Relevant  
Insight

# Modeling catalysis under operando conditions

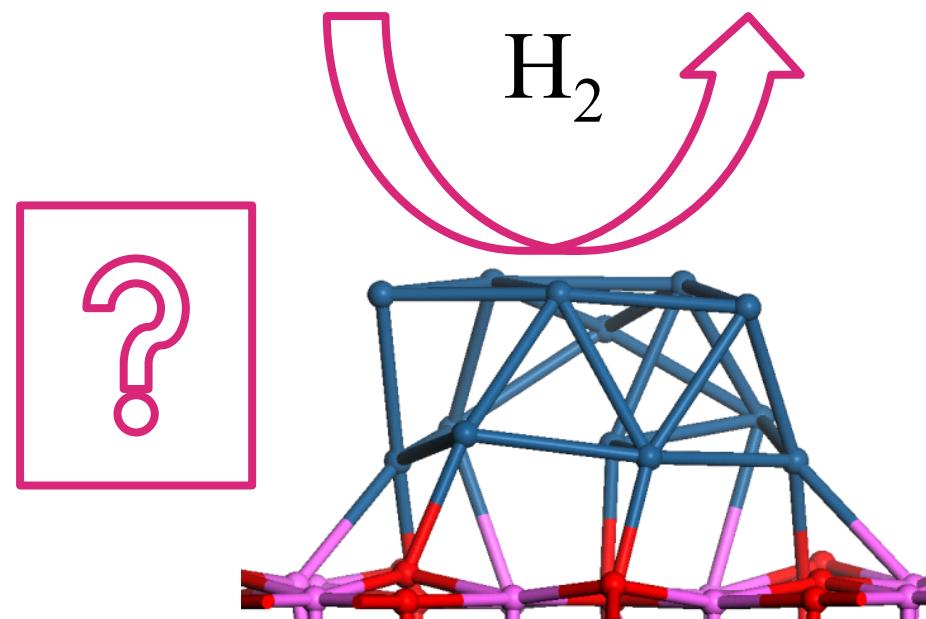
Conditions: T, P, rate, flow, liquid

Model: catalyst nature and geometry: slab, supported cluster  
Kinetic model, KMC lattice  
Reactor model

T, P : structure of catalyst in situ (Ab initio atomistic thermodynamics)  
Relation with experimental characterisation



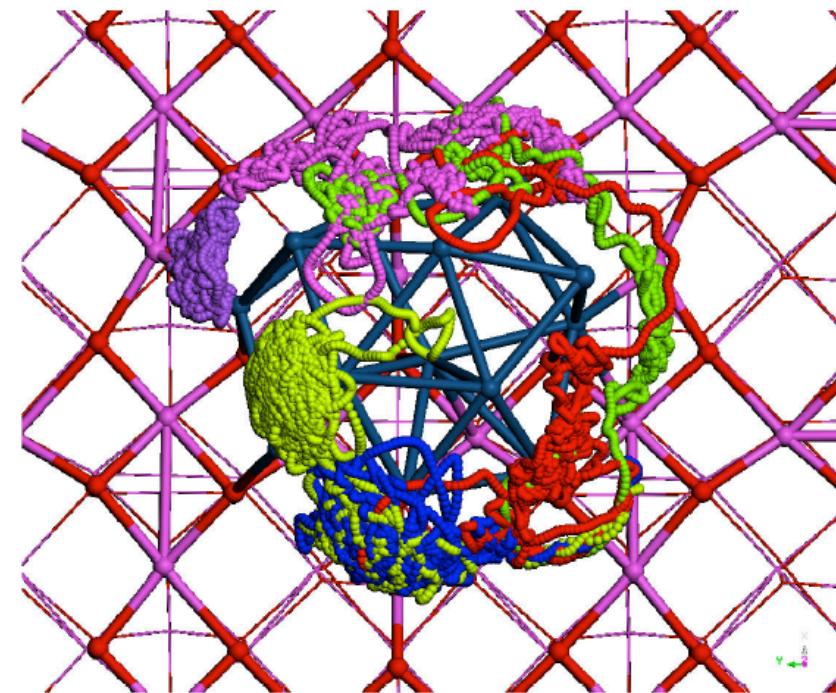
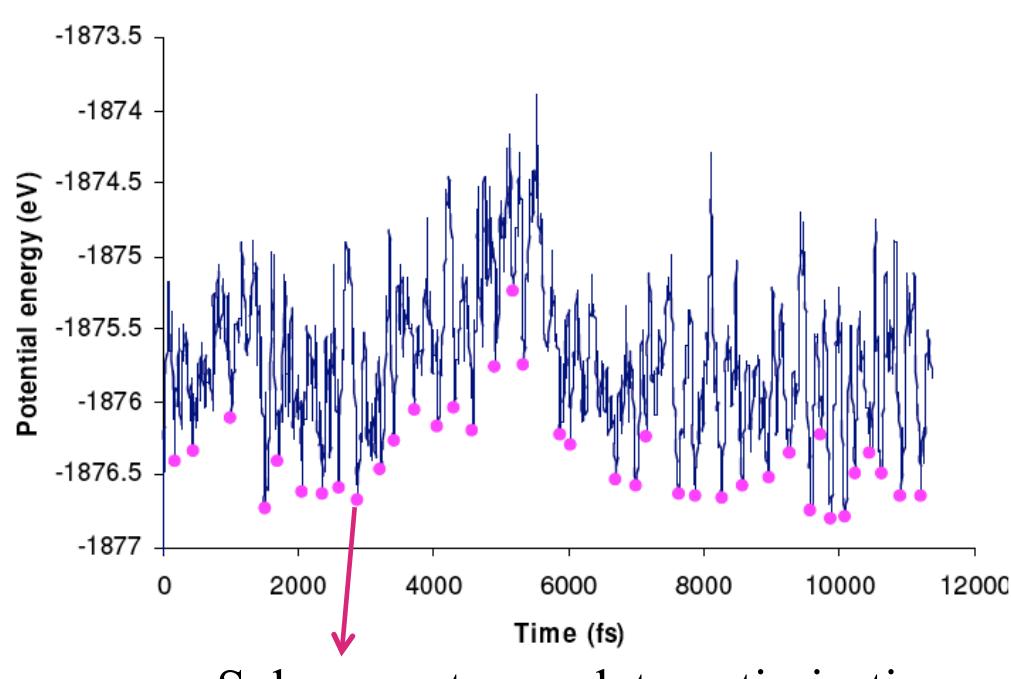
# $\text{Pt}_{13}$ on $\gamma\text{-Al}_2\text{O}_3$ under a pressure of $\text{H}_2$



Number of H atoms as a function of (P, T) ?



# $\text{Pt}_{13} + 6 \text{ H}$ on $\gamma\text{-Al}_2\text{O}_3$ (100)

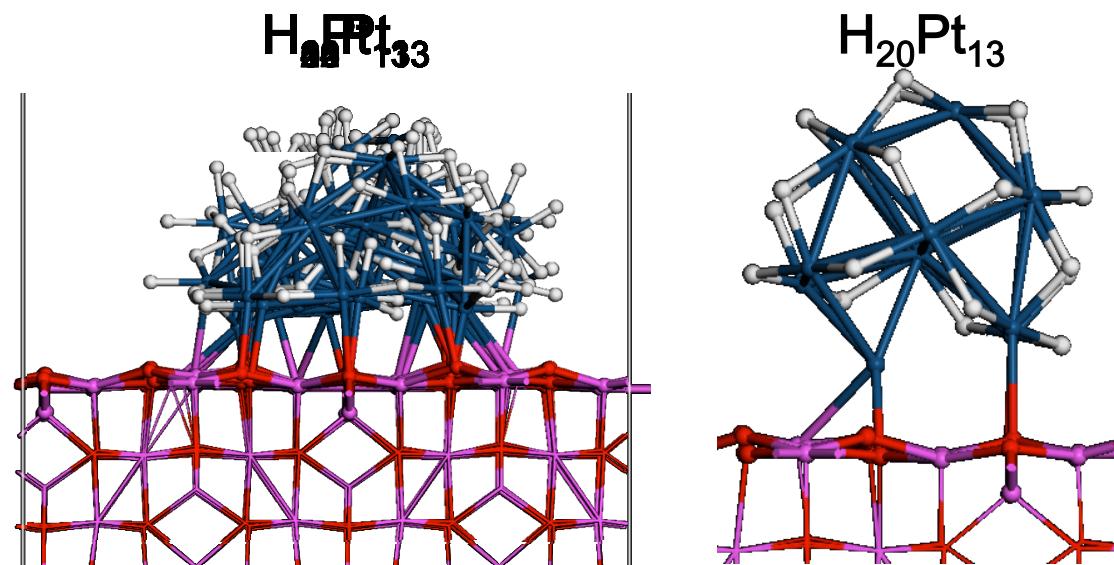


Velocity scaled MD, 1200 K, 12 ps,  $m_{\text{H}}=10$   
 $\text{Pt}_{13}$  and alumina frozen

C. Mager-Maury, C. Chizallet, P. Sautet, P. Raybaud ChemCatChem 3 (2011) 200



# Hydrogen adsorption: Pt<sub>13</sub>/(100) $\gamma$ -Al<sub>2</sub>O<sub>3</sub>

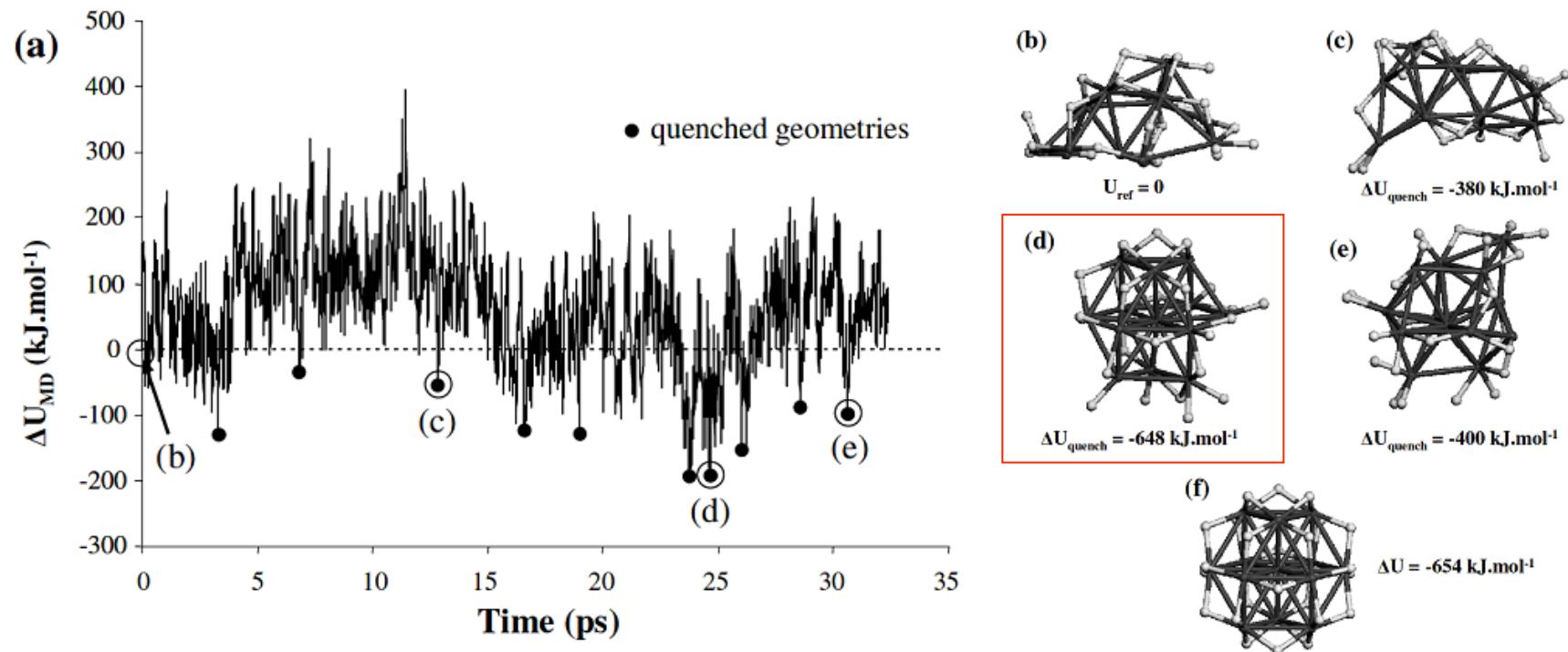


**Strong structural deformation of the Pt<sub>13</sub> cluster  
Weakening of the metal support interaction**

⇒ Change of the morphology under reductive environment  
⇒ Cuboctahedron is stabilized at high p(H<sub>2</sub>)

# Hydrogen adsorption: structural reconstruction

Gas phase molecular dynamic at n(H)=24

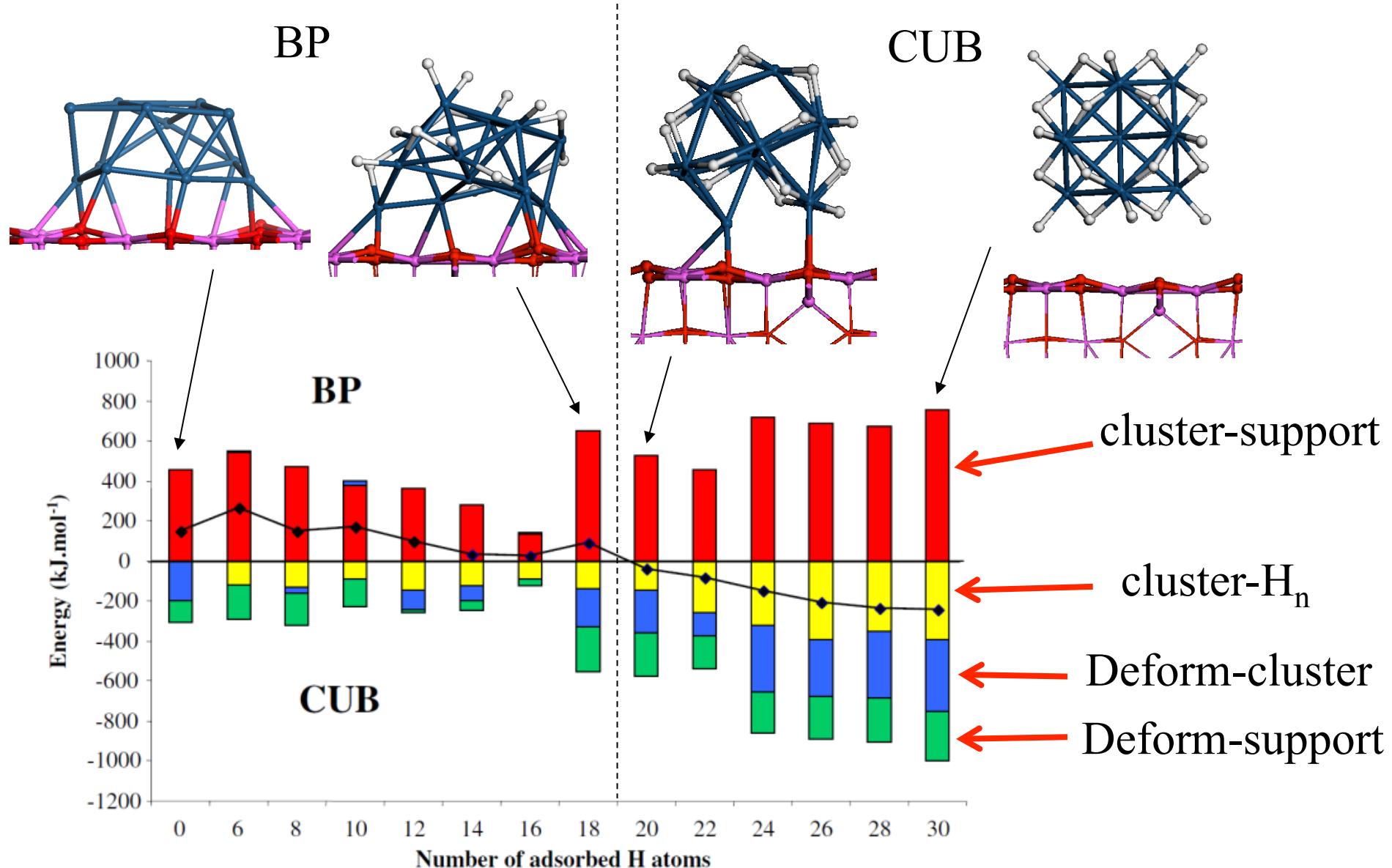


From BP to CUB transformation

C. Mager-Maury, C. Chizallet, P. Sautet, P. Raybaud ChemCatChem 3 (2011) 200

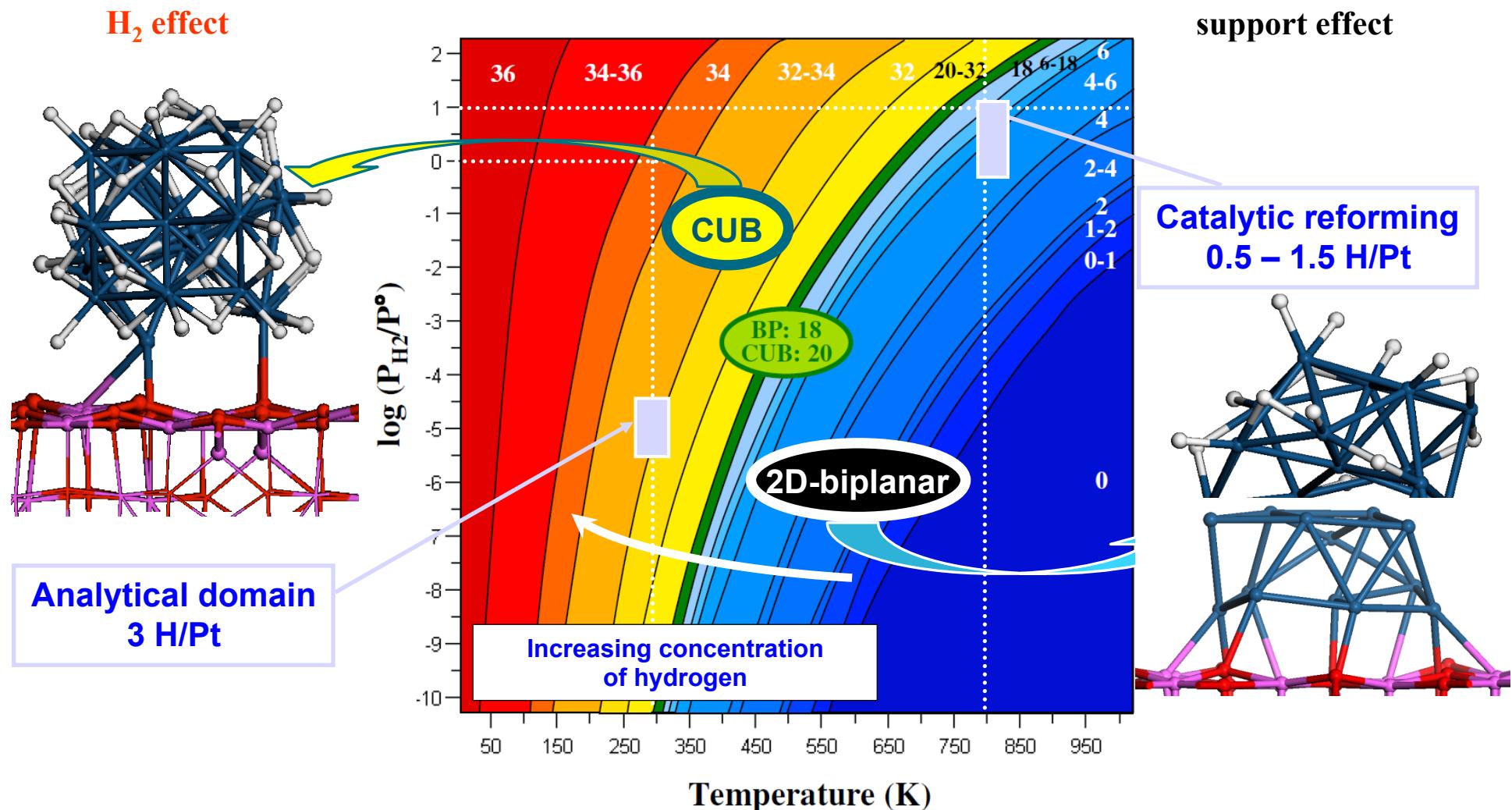


# Effect of Hydrogen: structural reconstruction



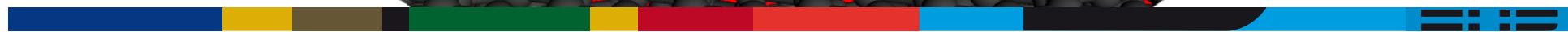
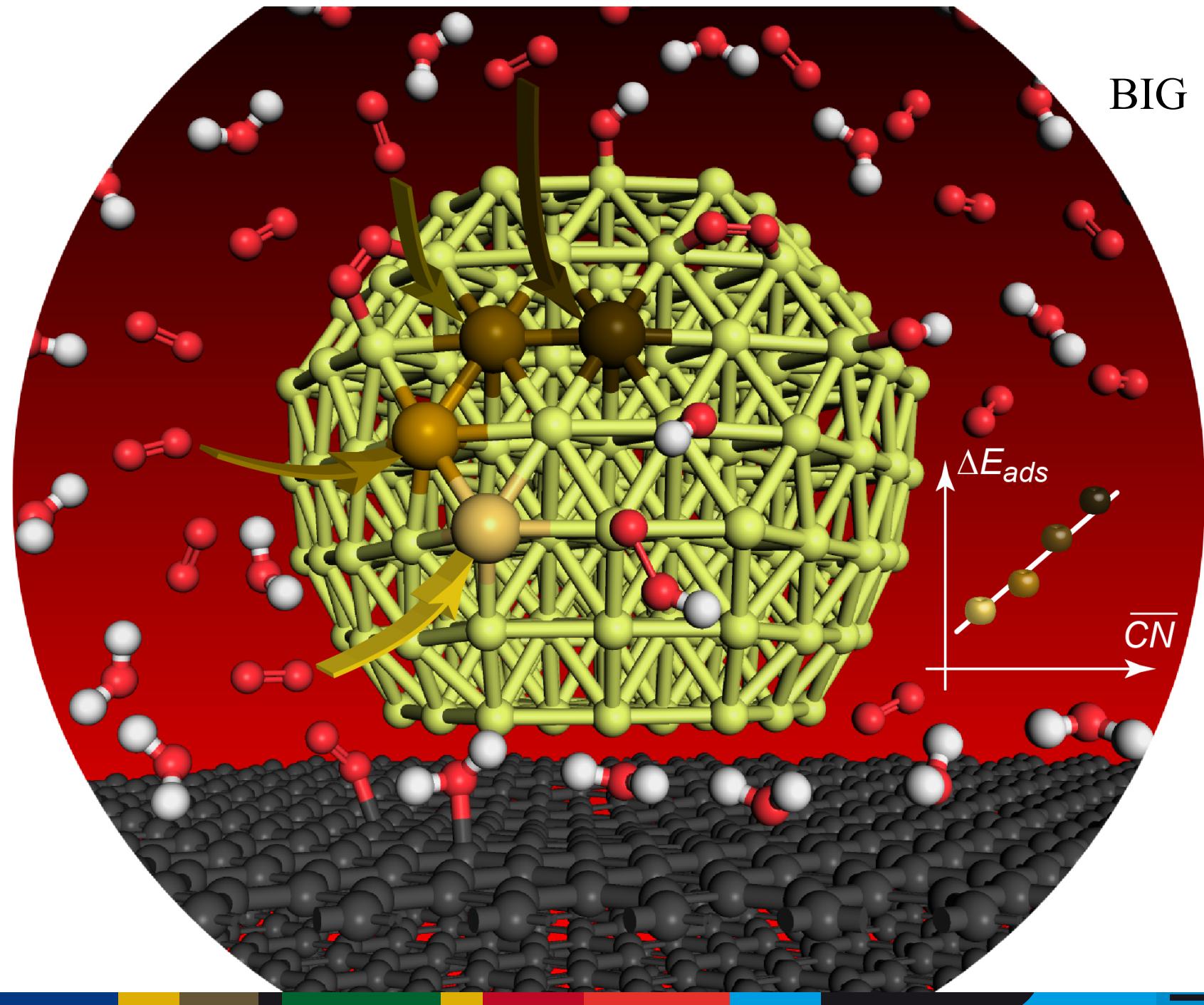
# Influence of H<sub>2</sub> on the structural properties of Pt<sub>13</sub>/γ-Al<sub>2</sub>O<sub>3</sub>-(100)

Phase diagram of Pt<sub>13</sub>-H<sub>n</sub>/(100) Al<sub>2</sub>O<sub>3</sub> as a function of T and P



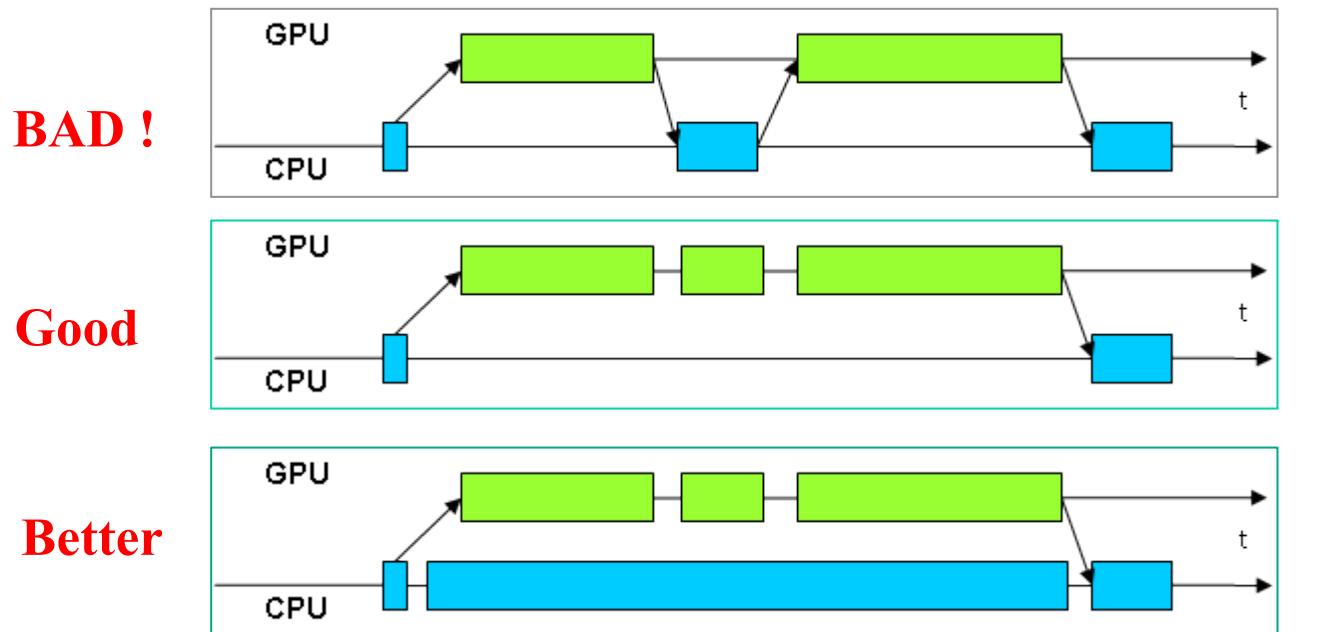
C. Mager-Maury, C. Chizallet, P. Sautet, P. Raybaud ChemCatChem 3 (2011) 200





# Porting VASP to GPU

- First step
  - FFTW -> CUFFT
  - BLAS -> CUBLAS
- Second step: minimizing data transfer time
  - By computing in parallel with CPU
  - By porting functions called between GPU calls



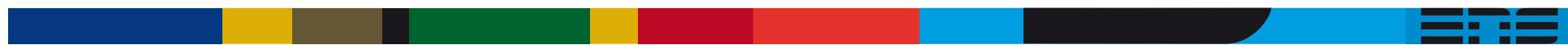
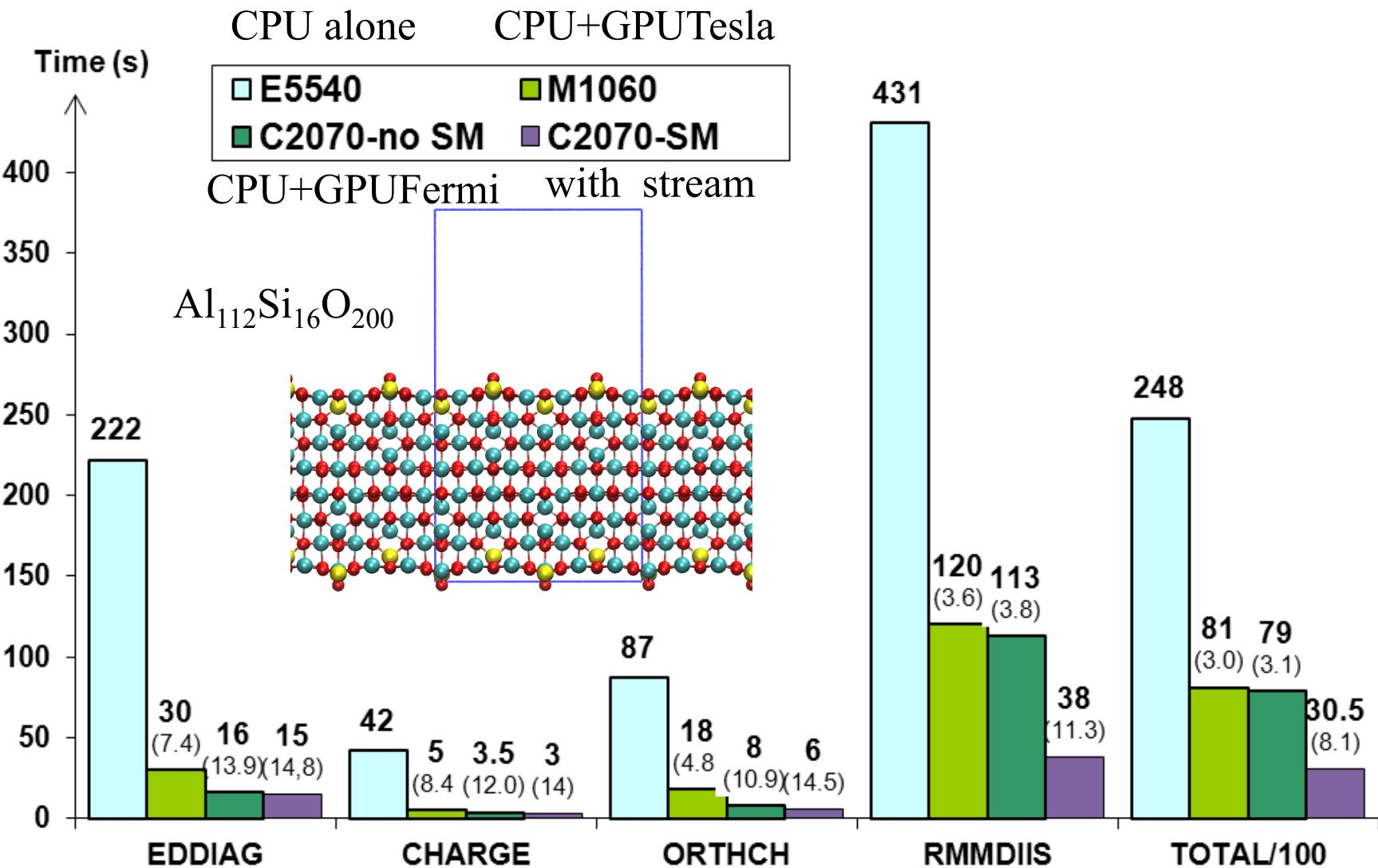
# Porting VASP to GPU

- First step
    - FFTW -> CUFFT
    - BLAS -> CUBLAS
  - Second step: minimizing data transfer time
    - By computing in parallel with CPU
    - By porting functions called between GPU calls
  - Third step: specific optimizations
    - EDDAV
    - EDDIAG and RMMDIIS
    - POTLOK, ORTHCH and CHARGE
- ⇒ Take home: Fill the GPU as much as possible using “streams”

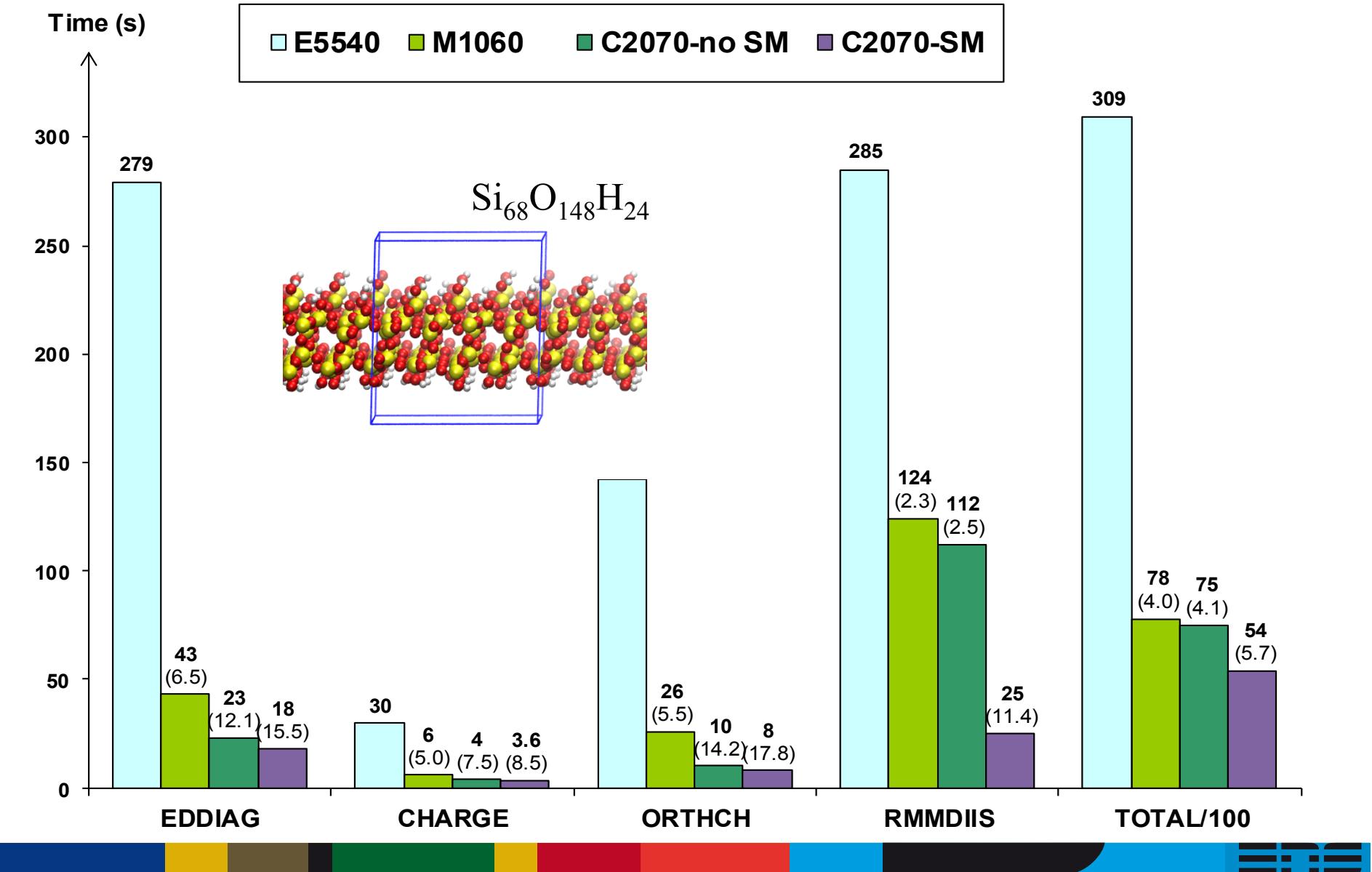
Hacene, M.; Anciaux-Sedrakian, A.; Rozanska, X.; Klahr, D.; Guignon T.; Fleurat-Lessard, P.  
J. Comput. Chem. (2012)



# Some results



# Some results



# Conclusion/Perspectives

- On penultimate generation of GPUs (Fermi):
  - No loss in scalability
  - Acceleration between 5.7 and 8.0
- Project with G. Kresse and Nvidia (Leader P. Fleurat-Lessard)
  - Merging our code with GPU exact exchange (M. Hutchinson, M. Widom)
  - Going to VASP 5.3, CUDA 5. to use fully the latest GPUs (K20)
  - Porting post-HF: MP2 and RPA

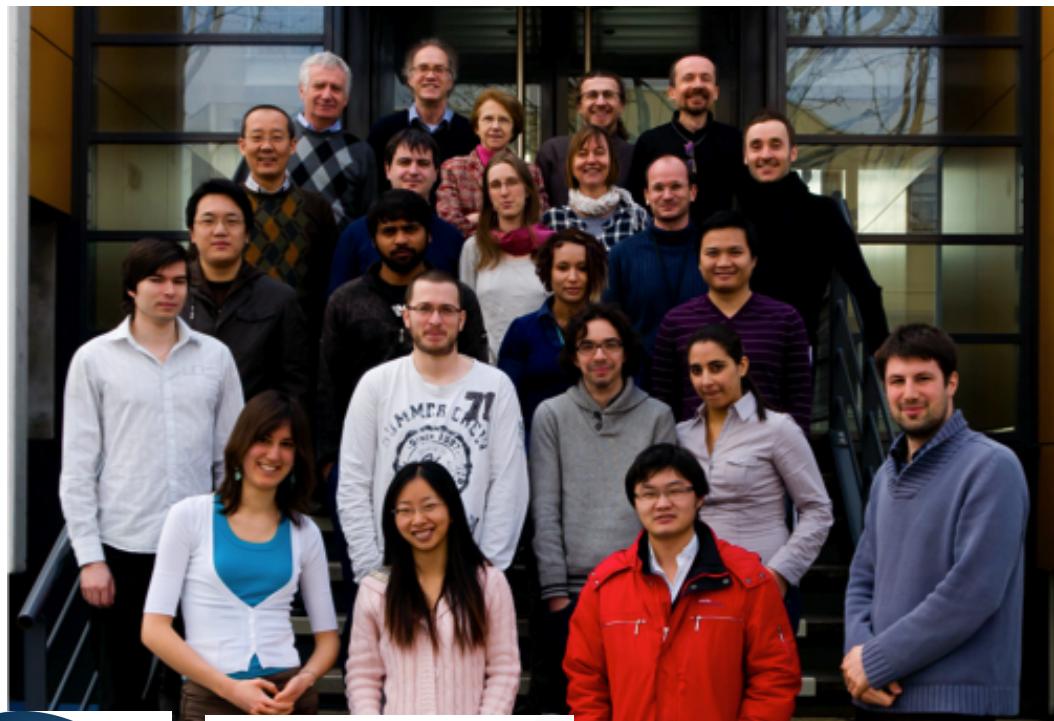
Hacene, M.; Anciaux-Sedrakian, A.; Rozanska, X.; Klahr, D.; Guignon T.; Fleurat-Lessard, P.  
J. Comput. Chem. (2012)



Lyon: F. Delbecq, C. Michel, D. Loffreda, M.L. Bocquet, P. Fleurat-Lessard, D. Torres, F. Cinquini, X. Rozanska, J. Zaffran, F. Auneau, C. Mager-Maury, R. Wischert, M. Iachella, F. Calle Valero, F. Görtl

IFPEN: P. Raybaud, C. Chizallet, M. Digne

Marne-La-Vallée: G. Chambaud, M. Guitou



Région Rhône-Alpes



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